

TABLE A1.  $P$ - $V$ - $T$  DATA FOR LITHIUM, SODIUM AND POTASSIUM(a) Lithium of natural isotopic composition ( $c = -1.1 \times 10^{-5} \text{ atm}^{-1}$ )

$T$ ( $^{\circ}\text{K}$ )	$V(T)/V(0)$	$-a$ ( $10^{-5} \text{ atm}^{-1}$ )	$-a'$ ( $10^{-5} \text{ atm}^{-1}$ )
0	1.000 <sub>0</sub>	0.77 <sub>6</sub>	0.77 <sub>6</sub>
50	1.000 <sub>9</sub>	0.77 <sub>7</sub>	0.77 <sub>7</sub>
100	1.003 <sub>5</sub>	0.78 <sub>1</sub>	0.78 <sub>0</sub>
150	1.007 <sub>7</sub>	0.79 <sub>5</sub>	0.79 <sub>0</sub>
200	1.013 <sub>1</sub>	0.82 <sub>2</sub>	0.80 <sub>5</sub>
250	1.019 <sub>4</sub>	0.85 <sub>6</sub>	0.83 <sub>0</sub>
300	1.026 <sub>7</sub>	0.89 <sub>4</sub>	0.86 <sub>0</sub>

(b) Sodium ( $c = -3.0 \times 10^{-5} \text{ atm}^{-1}$ )

0	1.000 <sub>0</sub>	1.40 <sub>0</sub>	1.40 <sub>0</sub>
50	1.001 <sub>4</sub>	1.40 <sub>6</sub>	1.40 <sub>0</sub>
100	1.006 <sub>6</sub>	1.42 <sub>4</sub>	1.40 <sub>0</sub>
150	1.014 <sub>9</sub>	1.46 <sub>3</sub>	1.40 <sub>0</sub>
200	1.024 <sub>3</sub>	1.50 <sub>7</sub>	1.40 <sub>0</sub>
250	1.034 <sub>2</sub>	1.55 <sub>2</sub>	1.40 <sub>0</sub>
300	1.044 <sub>2</sub>	1.60 <sub>0</sub>	1.40 <sub>0</sub>

(c) Potassium ( $c = -6.7 \times 10^{-5} \text{ atm}^{-1}$ )

0	1.000 <sub>0</sub>	2.80 <sub>0</sub>	2.80 <sub>0</sub>
50	1.002 <sub>3</sub>	2.83 <sub>0</sub>	2.80 <sub>5</sub>
100	1.011 <sub>9</sub>	2.95 <sub>0</sub>	2.82 <sub>0</sub>
150	1.023 <sub>4</sub>	3.08 <sub>3</sub>	2.83 <sub>0</sub>
200	1.034 <sub>7</sub>	3.22 <sub>2</sub>	2.83 <sub>0</sub>
250	1.046 <sub>0</sub>	3.35 <sub>5</sub>	2.82 <sub>8</sub>
300	1.057 <sub>3</sub>	3.49 <sub>0</sub>	2.82 <sub>0</sub>

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